

## A List of and Some Comments about the Trail Pheromones of Ants

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Ants use many different chemical compounds to communicate with their nestmates. Foraging success depends on how efficiently ants communicate the presence of food and thus recruit workers to exploit the food resource. Trail pheromones, produced by different exocrine glands, are a key part of ant foraging strategies. By combing through the literature, we compiled a list of the identity and glandular origin of the chemical compounds found in the trail pheromones of 75 different ant species. Of the 168 compounds identified, more than 40% are amines. In the subfamily Myrmicinae, trail pheromones are mostly produced in the venom gland, while in the subfamily Formicinae, they come from the rectal gland.

**Keywords:** Ants, Formicidae, Trail pheromone, Glands, Foraging strategies.

There are more than 12,700 known ant species in the world [1]. These species demonstrate many differences at both the colony (society) and individual (worker) level: colony size ranges from only a few workers to 20,000,000 individuals, and ants display a huge variety of foraging strategies that allow them to efficiently exploit their environment. The larger the colony, the less foraging is individually based and the more foraging is chemically coordinated [2].

Some ant species have lost the pheromone-based mechanisms that would allow them to communicate the location of a food resource. However, in some of these species, workers that have encountered a large food resource are still able to prompt their nestmates to exit the nest [3]. Yet, the vast majority of ant species do use chemicals to communicate the location of food and to recruit nestmates [2, 4]. Chemical recruitment mechanisms can be divided into three categories. In tandem running, a forager that has discovered food recruits a single nestmate at a time using an individual-specific trail [5]. As a result, this recruitment mechanism is the least efficient of the three. In group recruitment, a successful forager lays down a trail while returning to the nest and then guides a group of nestmates back to the food resource [6]. In mass recruitment, a successful forager lays down a chemical trail that elicits trail-following and trail-laying behavior by nestmates. This mechanism is the most efficient of the three since it does not require the successful worker to accompany the recruits to the resource. In this review, we broadly define trail pheromones and take into account all three recruitment mechanisms. For example, *Manica rubida* workers do not recruit nestmates but rather mark the ground with chemical trails upon their return to the nest after discovering prey [7]. *Aphaenogaster senilis* workers do the same but subsequently lead a small group of recruits along the trail to the food resource [6a]. Interest in studying and identifying ant trail pheromones has increased exponentially since the first trail pheromone was identified in 1971 [8]. However, trail pheromones have been studied

in a very small percentage of ant species because ants are so speciose (Table 1).

**Table 1:** Number of ant species identified which trail pheromones.

Subfamily	Total number of described species in the World <sup>a</sup>	Total number of species studied	% of total species studied
Myrmicinae (Myr)	6294	45	0.71
Dolichoderinae (Dol)	753	3	0.40
Formicinae (For)	3009	18	0.60
Ponerinae (Pon)	1087	5	0.46
Ectatomminae (Ect)	285	3	1.05
Dorylinae (Doryl)	59	1	1.70

<sup>a</sup>From [1] [http://osuc.biosci.ohio-state.edu/hymenoptera/tsa.sppcount?the\\_taxon=\[subfamily name\]](http://osuc.biosci.ohio-state.edu/hymenoptera/tsa.sppcount?the_taxon=[subfamily name]) (21st May 2014).

Some of these studies have been centered on ant species of economic interest, such as leaf-cutter ants (genus *Atta* or *Acromyrmex*), which are major herbivores that cause significant economic losses in agroecosystems, and invasive species, such as the red imported fire ant (*Solenopsis invicta*) and the Argentine ant (*Linepithema humile*). Studying ant pheromones has an applied interest because the knowledge gleaned can be used to help control and reduce the negative effects of these species. Some studies have used trail or alarm pheromones to increase ants' consumption of toxic baits [9]. Other studies have focused on the use of natural or synthetic semiochemicals to disrupt ant trails [10]. Recently, the success of trail pheromone disruption trials in a natural ecosystem (Volcanoes National Park in Hawaii) has underscored the potential that these methods have to control invasive ant species [10c].

Over the last two decades, different publications have reviewed the pheromones found in social insects [11]. Moreover, two excellent reviews have focused specifically on ant trail pheromones [12]: the most recent review [12b] described 54 such compounds. We have added to this list, bringing the total up to 75. In this review, we provide a list of chemical compounds that function as trail pheromones (in a broader sense of the term) and their glandular origin.

Trail pheromones are produced by either one or a few glands located in the gaster, such as the venom gland (in the subfamily Myrmicinae) and the rectal gland (in the subfamily Formicinae) (Table 2). The Dufour, pygidial, post-pygidial, rectal, sternal, and Pavan glands, also found in the abdomen, are other sources of trail pheromones. *Crematogaster castanea* workers are unable to touch the ground with the tip of their heart-shaped abdomens: they use a secretion from their tibial glands to lay down trails [13]. In *Leptogenys diminuta*, workers lay down trails composed of secretions from both the poison and the pygidial glands [14]. The poison gland secretions provide the orientation cues while those from the pygidial gland stimulate recruitment [14]. Trail pheromones generally comprise several compounds (Table 3, Figure 1). The chemical blend is highly variable among species.

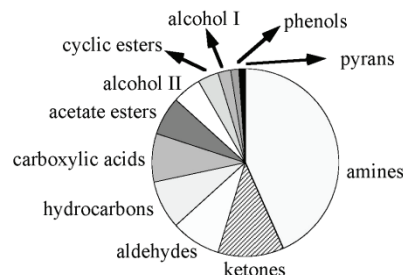
**Table 2:** Number of ant species in the different subfamilies<sup>a</sup> that use chemical compounds from various glands as trail pheromones.

Gland	Myr	Dol	Form	Pon	Ect	Dor
venom (V)	36			1		
Dufour (D)	1				2	
pygidial (Py)	2	1		1		
rectal (R)			15			
sternal (S)						
postpygidial (Pp)						1
tibial (T)	1					
Pavan (Pa)		1				
V+D	4					
V+Py				2		
V+R			1			
Py+Pa		1				

<sup>a</sup> See Table 1 for taxon abbreviations

For example, in *L. diminuta*, a single compound [(3*R*,4*S*)-4-methylheptan-3-ol] is enough to provoke trail-following behavior, whereas in *L. pequeti*, the trail pheromone is composed of 14 compounds; each of them is individually able to elicit some trail-following behavior, but the mixture of all the substances is more effective than any single compound [15]. Closely related species with morphologically similar workers may nonetheless have different trail pheromones. For example, *Tetramorium caespitum* uses a mixture of the pyrazines (or tetramorines) DMP and EDMP, while *T. impurum* uses methyl 6-methylsalicylate [16]. In the army ant *Daceton armigerum*, there are two types of trail pheromone compounds: a mixture of linear alkanes and alkenes secreted by the Dufour gland and a mixture of tetramorines (DMP, TMP, and EDMP) secreted by the poison gland [17]. The linear hydrocarbons might be acting as solvents, "retaining" the more volatile tetramorines [17]. Similarly, the poison gland secretion of the predatory ant *Myrmecaria eumenoides* has a high volatile fraction—which is mostly composed of (+)-limonene, the recruitment pheromone—and a low volatile fraction whose main component is an alkaloid that acts as a fixative and modifies limonene evaporation kinetics, extending the period over which the signal is effective [18]. Chemical communication involves a short-lived excitation signal combined with a long-lasting orientation signal [19]. Some species might also use repellent pheromones to prevent positive feedback when necessary [20]. The complex trail system of *Paratrechina longicornis* demonstrates extraordinary adaptability and integrates signals of different strength and duration: stable routes to either permanent food resources or nest sites are maintained with rectal pheromones, while temporary recruitment trails to unstable but profitable resources are generated with either Dufour or poison gland pheromones [21].

The longevity of trail pheromone signals is highly variable among species. In the army ant *Daceton armigerum*, trails created with poison gland secretions can last for more than seven days [22]. In the Pharaoh's ant, *M. pharaonis*, trail decay is rapid (pheromone half-life is 9 min) and is strongly affected by the substrate upon



**Figure 1:** Chemical compounds used as trail pheromones by ants. Data obtained from literature (168 compounds from 75 ant species). See Table 3 for details

which the trail is laid [23]. In this species, there are three types of trail pheromone: a long-lasting attractive pheromone and two short-lived pheromones, one attractive and one repellent [20b]. The use of long-lasting compounds in trails allows workers to re-establish trails after 48 h have passed [24].

Another factor that could affect the longevity of trail pheromone signals is ground surface temperature. Hence, in some Mediterranean environments, hot ground-surface temperatures may accelerate trail pheromone evaporation and thus recruitment decay [25]. Microclimate might therefore explain why many species living in such habitats use individual foraging and lack recruitment pheromones [26]. On the other hand, thermophilic ant species are mainly scavengers, exploiting unpredictable food resources with patchy distributions. Consequently, trail-based worker recruitment may not be advantageous. For example, in contrast to its sister species from North America, the South American seed-harvester ant *Pogonomyrmex vermiculatus* is a solitary forager, probably because it exploits a patchy food resource (seeds occur in low densities in the Chilean desert) in a low-competition environment [27].

**Perspectives**—Trail pheromone compounds have always been identified indirectly: they are extracted from their glandular sources and bioassays are performed using the extracts or synthesized standards and their mixtures [28]. As Cross *et al.* [29] pointed out: "It is an oversimplification to ascribe the complex social behavior of trail following solely to the compound identified as the most active, satisfying though it may be to elicit this response in the laboratory and in the field with a defined chemical". Very recently, Choe *et al.* [28] highlighted a methodological problem concerning the identification of one ant trail pheromone: for more than three decades, different studies have stated that (*Z*)-9-hexadecenal might be a key component of Argentine ant trails because it strongly attracted workers in a multi-choice olfactometer. Using a very original experimental set-up, Choe *et al.* [28] directly studied the substances deposited by living ants as they created their trails. They collected trail chemicals by providing ants with solid-phase microextraction fibers as bridges during foraging (as ants moved from the nest to the feeder). They found that (*Z*)-9-hexadecenal is not present in detectable quantities and that two iridoids, dolichodial and iridomyrmecin, appear to be the primary chemical constituents of the trails. We think that Choe *et al.*'s [28] work presents a novel and fruitful approach to study ant trail pheromones: by obtaining them directly from the trail.

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**Table 3:** Chemical compounds identified as trail pheromones in the different ant species.

Ant species	Chemical compounds	Gland	Functional group	Chain length	Molecular weight	[reference]
<b>Subfamily MYRMICINAE</b>						
<b>Tribe Dacetini</b>						
<i>Daceton armigerum</i>	2,5-dimethylpyrazine (DMP)	venom	Amine	C03	108.14	[17]
	2,3,5-trimethylpyrazine (TMP)	venom	Amine	C04	122.17	[17]
	3-ethyl-2,5-dimethylpyrazine (EDMP)	venom	Amine	C05	136.19	[17]
	9-tricosene	Dufour	Hydrocarbon	C23	324.63	[17]
	tricosane	Dufour	Hydrocarbon	C23	324.63	[17]
	9-pentacosene	Dufour	Hydrocarbon	C25	350.66	[17]
<b>Tribe Solenopsidini</b>						
<i>Mayriella overbecki</i>	methyl 6-methylsalicylate	venom	Phenol	C08	166.17	[30]
<i>Monomorium pharaonis</i>	3-butyl-5-methyloctahydroindolizine (monomorine I)	venom	Amine	C13	195.34	[31]
	2(5'-hexenyl)-5-pentyl pyrrolidine (monomorine III)	venom	Amine	C15	223.4	[32]
	(6E,10Z)-3,4,7,11-tetramethyl-6,10-tridecadien-1-al (faranal)	Dufour	Carboxylic acid	C13	250.42	[33]
<i>Solenopsis invicta</i>	Z,E-alpha-farnesene	Dufour	Hydrocarbon	C12	204.35	[34]
	Z,E-alpha-homofarnesene	Dufour	Hydrocarbon	C12	218.38	[35]
	Z,Z-alpha-homofarnesene	Dufour	Hydrocarbon	C12	218.38	[35]
	Z,Z,Z-allofarnesene	Dufour	Hydrocarbon	C12	204.35	[36]
<b>Tribe Myrmicini</b>						
<i>Eutetramorium mocquersyi</i>	2,3-dimethyl-5-(2-methylpropyl)-pyrazine	venom	Amine	C05	164.25	[37]
<i>Manica rubida</i>	EDMP	venom	Amine	C05	136.19	[38]
<i>Myrmica lobicornis</i>	EDMP	venom	Amine	C05	136.19	[39]
<i>Myrmica rubra</i>	EDMP	venom	Amine	C05	136.19	[39]
<i>Myrmica ruginodis</i>	EDMP	venom	Amine	C05	136.19	[39]
<i>Myrmica rugulosa</i>	EDMP	venom	Amine	C05	136.19	[39]
<i>Myrmica sabuleti</i>	EDMP	venom	Amine	C05	136.19	[39]
<i>Myrmica scabrinodis</i>	EDMP	venom	Amine	C05	136.19	[39]
<i>Myrmica schencki</i>	EDMP	venom	Amine	C05	136.19	[39]
<i>Myrmica sulcinodis</i>	EDMP	venom	Amine	C05	136.19	[39]
<i>Pogonomyrmex barbatus</i>	TMP	venom	Amine	C04	122.17	[40]
	EDMP	venom	Amine	C05	136.19	[40]
	DMP	venom	Amine	C03	108.14	[40]
<i>Pogonomyrmex maricopa</i>	TMP	venom	Amine	C04	122.17	[40]
	EDMP	venom	Amine	C05	136.19	[40]
	DMP	venom	Amine	C03	108.14	[40]
<i>Pogonomyrmex occidentalis</i>	TMP	venom	Amine	C04	122.17	[40]
	EDMP	venom	Amine	C05	136.19	[40]
	DMP	venom	Amine	C03	108.14	[40]
<i>Pogonomyrmex rugosus</i>	TMP	venom	Amine	C04	122.17	[40]
	EDMP	venom	Amine	C05	136.19	[40]
	DMP	venom	Amine	C03	108.14	[40]
<b>Tribe Tetramorini</b>						
<i>Tetramorium caespitum</i>	DMP	venom	Amine	C03	108.14	[16]
	EDMP	venom	Amine	C05	136.19	[16]
<i>Tetramorium impurum</i>	methyl 6-methylsalicylate	venom	Phenol	C08	166.17	[16c, 41]
<i>Tetramorium meridionale</i>	2-methylpyrazine (MP)	venom	Amine	C03	94.11	[42]
	DMP	venom	Amine	C03	108.14	[42]
	TMP	venom	Amine	C04	122.17	[42]
	EDMP	venom	Amine	C05	136.19	[42]
	1-benzo[b]pyrrole (indole)	venom	Amine	C08	117.15	[42]
<b>Tribe Pheidolini</b>						
<i>Aphaenogaster albisetosus</i>	(S)-4-methyl-3-heptanone	venom	Ketone	C07	128.21	[43]
	(R)-4-methyl-3-heptanone	venom	Ketone	C07	128.21	[43]
<i>Aphaenogaster cockerelli</i>	(R)-1-phenylethanol	venom	Alcohol I	C08	122.16	[43]
	(S)-4-methyl-3-heptanone	venom	Ketone	C07	128.21	[43]
<i>Aphaenogaster rudis</i>	3-(2-piperidiny)pyridine (anabasine)	venom	Amine	C09	162.23	[44]
	3,4,5,6-tetrahydro-2,3'-bipyridine (anabaseine)	venom	Amine	C09	160.22	[44]
	2,3'-bipyridyl	venom	Amine	C09	156.18	[44]
	N-isopentyl-2-phenylethylamine	venom	Amine	C08	191.31	[44]
<i>Aphaenogaster senilis</i>	anabaseine	venom	Amine	C09	160.22	[45]
	anabaseine	venom	Amine	C09	162.23	[45]
	2,3'-bipyridine	venom	Amine	C09	156.18	[45]
<i>Messor andrei</i>	n-tridecane	pygidial	Hydrocarbon	C13	184.36	[46]
<i>Messor bouvieri</i>	anabaseine	venom	Amine	C09	162.23	[47]
	anabaseine	venom	Amine	C09	160.22	[47]
	EDMP	venom	Amine	C05	136.19	[47]
	heptadecadiene	Dufour	Hydrocarbon	C17	236.44	[48]
<i>Messor capensis</i>	anabaseine	venom	Amine	C09	162.23	[49]
	anabaseine	venom	Amine	C09	160.22	[49]
	anabaseine	venom	Amine	C09	162.23	[50]
<i>Messor ebeninus</i>	1-pentadecene	Dufour	Hydrocarbon	C15	210.4	[50]
<i>Messor pergandei</i>	n-tridecane	pygidial	Hydrocarbon	C13	184.36	[46]
<i>Pheidole pallidula</i>	EDMP	-	Amine	C05	136.19	[51]
<i>Pristomyrmex pungens</i>	6-n-pentyl-2-pyrone	venom	Ketone	C10	166.22	[52]

<b>Tribe Crematogastrini</b>						
<i>Crematogaster castanea</i>	(R)-2-dodecanol	tibial	Alcohol II	C12	186.33	[13]
<b>Tribe Attini</b>						
<i>Acromyrmex octospinosus</i>	methyl 4-methylpyrrole-2-carboxylate	venom	Amine	C05	139.15	[29, 53]
	EDMP	venom	Amine	C05	136.19	[29, 53]
	2-ethyl 3,5-dimethyl-pyrazine	venom	Amine	C05	136.2	[29, 53]
	acetaldehyde	venom	Aldehyde	C02	44.05	[53]
<i>Acromyrmex subterraneus molestans</i>	methyl 4-methylpyrrole-2-carboxylate	venom	Amine	C05	139.15	[54]
<i>Acromyrmex subterraneus subterraneus</i>	methyl 4-methylpyrrole-2-carboxylate	venom	Amine	C05	139.15	[55]
<i>Atta bisphaerica</i>	methyl 4-methylpyrrole-2-carboxylate	venom	Amine	C05	139.15	[54, 56]
	EDMP	venom	Amine	C05	136.19	[54, 56]
	2-phenylacetic acid	venom	Carboxylic acid	C08	136.15	[56]
	bornylene	venom	Hydrocarbon	C09	136.23	[56]
	1-octanol	venom	Alcohol I	C08	130.23	[56]
<i>Atta cephalotes</i>	methyl 4-methylpyrrole-2-carboxylate	venom	Amine	C05	139.15	[53, 57]
	EDMP	venom	Amine	C05	136.19	[57]
	acetaldehyde	venom	Aldehyde	C02	44.05	[57]
	acetone	venom	Ketone	C03	58.08	[57]
	methylpropanal	venom	Aldehyde	C04	72.11	[57]
<i>Atta laevigata</i>	methyl 4-methylpyrrole-2-carboxylate	venom	Amine	C05	139.15	[54, 56]
	2-phenylacetic acid	venom	Carboxylic acid	C08	136.15	[54, 56]
	bornylene	venom	Hydrocarbon	C09	136.23	[56]
<i>Atta sexdens</i>	methyl 4-methylpyrrole-2-carboxylate	venom	Amine	C05	139.15	[9a]
<i>Atta sexdens rubropilosa</i>	methyl 4-methylpyrrole-2-carboxylate	venom	Amine	C05	139.15	[53, 58]
	EDMP	venom	Amine	C05	136.19	[53, 58]
	acetaldehyde	venom	Aldehyde	C02	44.05	[53]
	acetone	venom	Ketone	C03	58.08	[53]
	methylpropanal	venom	Aldehyde	C04	72.11	[53]
	methyl 2-phenylacetate	venom	Acetate ester	C08	150.17	[58]
	ethyl 2-phenylacetate	venom	Acetate ester	C08	164.2	[58]
<i>Atta sexdens sexdens</i>	methyl 4-methylpyrrole-2-carboxylate	venom	Amine	C05	139.15	[53, 59]
	EDMP	venom	Amine	C05	136.19	[53, 59a]
	DMP	venom	Amine	C03	108.14	[59b]
	TMP	venom	Amine	C04	122.17	[59b]
	acetaldehyde	venom	Aldehyde	C02	44.05	[53, 59b]
	acetone	venom	Ketone	C03	58.08	[53]
	methylpropanal	venom	Aldehyde	C04	72.11	[53]
<i>Atta texana</i>	methyl 4-methylpyrrole-2-carboxylate	-	Amine	C05	139.15	[8, 60]
<b>Tribe Metoponini</b>						
<i>Metapone madagascariensis</i>	methyl pyrrole-2-carboxylate	venom	Amine	C05	125.13	[61]
	EDMP	venom	Amine	C05	136.19	[61]
	DMP	venom	Amine	C03	108.14	[61]
	TMP	venom	Amine	C04	122.17	[61]
	acetophenone	venom	Ketone	C08	120.15	[61]
<b>Tribe Myrmicariini</b>						
<i>Myrmecaria eumenoides</i>	(+)- limonene	venom	Hydrocarbon	C08	136.23	[18]
<b>Subfamily DOLICHODERINAE</b>						
<b>Tribe Dolichoderini</b>						
<i>Dolichoderus thoracicus</i>	(Z)-9-octadecenal	Pavan	Aldehyde	C18	266.47	[62]
	(Z)-9-hexadecenal	Pavan	Aldehyde	C16	238.41	[62]
	(Z)-11-eicosenal	Pavan	Aldehyde	C20	294.52	[62]
	(Z)-13-docosenal	Pavan	Aldehyde	C22	322.57	[62]
<b>Tribe Tapinomini</b>						
<i>Linepithema humile</i>	(Z)-9-hexadecenal	(ventral)	Aldehyde	C16	238.41	[63]
	dolichodial [=2-(1-formylvinyl)-5-methylcyclopentanecarbaldehyde]	pygidial	Aldehyde	C07	166.22	[28]
	iridomyrmecin [=hexahydro-4,7-dimethylcyclopenta[c]pyran-3(1H)-one]	pygidial	Ketone	C08	168.23	[28]
<i>Tapinoma simrothi</i>	iridodial [=2-acetyl-5-methylcyclopentanecarbaldehyde]	pygidial	Aldehyde	C07	154.21	[64]
	iridomyrmecin	pygidial	Ketone	C08	168.23	[64]
<b>Subfamily FORMICINAE</b>						
<b>Tribe Formicini</b>						
<i>Formica rufa</i>	mellein [3,4-dihydro-8-hydroxy-3-methylisocoumarin]	rectal	Ketone	C10	178.19	[65]
<b>Tribe Lasiini</b>						
<i>Lasius fuliginosus</i>	mellein	(hindgut)	Ketone	C10	178.19	[66]
	2,3-dihydro-3,5-dihydroxy-6-methyl-4(H)-pyran-4-one	(hindgut)	Ketone	C06	144.13	[66]
	hexanoic acid	(rectal fluid)	Carboxylic acid	C06	116.16	[67]
	heptanoic acid	(rectal fluid)	Carboxylic acid	C07	130.19	[67]
	octanoic acid	(rectal fluid)	Carboxylic acid	C08	144.21	[67]
	nonanoic acid	(rectal fluid)	Carboxylic acid	C09	158.24	[67]
	decanoic acid	(rectal fluid)	Carboxylic acid	C10	172.26	[67]
	dodecanoic acid	(rectal fluid)	Carboxylic acid	C12	200.32	[67]
<i>Lasius niger</i>	3,4-dihydro-8-hydroxy-3,5,7-trimethylisocoumarin	rectal	Ketone	C10	206.24	[65]
<b>Tribe Camponotini</b>						
<i>Camponotus atriceps</i>	6-sec-butyl-tetrahydro-3,5-	rectal	Pyran	C08	184.26	[68]

<i>Camponotus balzani</i>	dimethylpyran-2-one 3,4-dihydro-8-hydroxy-3,5,7-trimethylisocoumarin	rectal	Ketone	C08	206.24	[69]
<i>Camponotus castaneus</i>	3,5-dimethyl-6-(1'-methylpropyl)-tetrahydro-2H-pyran-2-one	rectal	Pyran	C08	184.26	[69]
<i>Camponotus floridanus</i>	nerolic acid [= (Z)-3,7-dimethyl-2,6-octadienoic]	rectal	Carboxylic acid	C08	168.23	[68]
<i>Camponotus herculeanus</i>	(2S,4R,5S)-2,4-dimethyl-5-hexanolide	rectal	Cyclic ester	C06	142.2	[70]
<i>Camponotus inaequalis</i>	3,4-dihydro-8-hydroxy-3,5,7-trimethylisocoumarin	rectal	Ketone	C10	206.24	[71]
<i>Camponotus ligniperda</i>	(2S,4R,5S)-2,4-dimethyl-5-hexanolide	rectal	Cyclic ester	C06	142.2	[70b]
<i>Camponotus mus</i>	5-hydroxy-4-decanolide	-	Cyclic ester	C10	186.25	[72]
<i>Camponotus pennsylvanicus</i>	(2S,4R,5S)-2,4-dimethyl-5-hexanolide	rectal	Cyclic ester	C06	142.2	[70b]
<i>Camponotus planatus</i>	5-hydroxy-4-decanolide	-	Cyclic ester	C10	186.25	[72]
<i>Camponotus rufipes</i>	3,4-dihydro-8-hydroxy-3,7-dimethylisocoumarin	rectal	Ketone	C10	178.19	[73]
<i>Camponotus sericeiventris</i>	3,4-dihydro-8-hydroxy-3,5,7-trimethylisocoumarin	rectal	Ketone	C08	206.24	[69]
<i>Camponotus silvicola</i>	3,4-dihydro-8-hydroxy-3,5,7-trimethylisocoumarin	rectal	Ketone	C08	206.24	[73]
<i>Camponotus socius</i>	(2S,4R,5S)-2,4-dimethyl-5-hexanolide	rectal	Cyclic ester	C06	142.2	[70b]
	2,3-dihydro-3,5-dihydroxy-6-methyl-4(H)-pyran-4-one	rectal	Ketone	C06	144.13	[74]
	formic acid [=methanoic acid]	venom	Carboxylic acid	C01	46.03	[74]
<i>Camponotus vagus</i>	(2S,4R,5S)-2,4-dimethyl-5-hexanolide	rectal	Cyclic ester	C06	142.2	[70b]
<b>Subfamily PONERINAE</b>						
<b>Tribe Ponerini</b>						
<i>Leptogenys diminuta</i>	(3R,4S)-4-methyl-3-heptanol	venom	Alcohol II	C07	130.23	[14, 75]
	cis-isogeraniol	pygidial	Alcohol I	C08	154.25	[14b, 75]
<i>Leptogenys peuqueti</i>	1-ethyl-4-methylheptyl acetate	venom	Acetate ester	C08	200.32	[15]
	1-isopropyl-4-methylheptyl acetate	venom	Acetate ester	C08	214.34	[15]
	1-propyl-4-methylheptyl acetate	venom	Acetate ester	C09	214.34	[15]
	4-methyl-7-dodecanol	venom	Alcohol II	C11	200.36	[15]
	3,9-dimethyl-6-dodecanol	venom	Alcohol II	C12	214.39	[15]
	1-pentyl-4-methylheptyl acetate	venom	Acetate ester	C11	242.4	[15]
	4-methyl-7-tridecanol	venom	Alcohol II	C13	214.39	[15]
	4-methyl-7-tetradecanol	venom	Alcohol II	C14	228.41	[15]
	4,10-dimethyl-7-tridecanol	venom	Alcohol II	C13	228.41	[15]
	1-(3-methylhexyl)-4-methylheptyl acetate	venom	Acetate ester	C13	270.45	[15]
	1-(3-methylhexyl)-octyl acetate	venom	Acetate ester	C14	270.45	[15]
	1-heptyloctyl acetate	venom	Acetate ester	C15	270.45	[15]
	4-methyl-7-hexadecanol	venom	Alcohol II	C16	256.47	[15]
	1-(3-methylhexyl)-decyl acetate	venom	Acetate ester	C16	298.5	[15]
<i>Megaponera foetens</i> (= <i>Pachycondyla analis</i> )	N,N-dimethyluracil	venom	Amine	C03	140.14	[76]
	actinidine	pygidial	Amine	C08	147.22	[76b]
<i>Pachycondyla marginata</i>	citronellal [=3,7-dimethyl-6-octenal]	pygidial	Aldehyde	C08	154.25	[77]
<i>Pachycondyla tarsata</i>	9-heptadecanone	sternal	Ketone	C17	254.45	[78]
<b>Subfamily ECTATOMMINAE</b>						
<b>Tribe Ectatommini</b>						
<i>Gnamptogenys striatula</i>	4-methylgeranyl decanoate	Dufour	Carboxylic ester	C10	322.53	[79]
	4-methylgeranyl dodecanoate	Dufour	Carboxylic ester	C12	350.58	[79]
<i>Ectatomma ruidum</i>	geranylgeraniol acetate	Dufour	Acetate ester	C16	332.52	[80]
	geranylgeraniol	Dufour	Alcohol I	C16	290.48	[80]
<i>Rhytidoponera metallica</i>	isogeraniol [= (Z)-3,7-dimethyl-3,6-octadien-1-ol]	pygidial	Alcohol I	C08	154.25	[81]
	3-hydroxybenzaldehyde	pygidial	Aldehyde	C06	122.12	[81]
<b>Subfamily DORYLINAE</b>						
<b>Tribe Aenictini</b>						
<i>Aenictus sp</i>	methyl anthranilate [=methyl 2-aminobenzoate]		Carboxylic ester	C07	151.16	[82]
	methyl nicotinate [=methyl 3-pyridinecarboxylate]		Amine	C05	137.14	[82]

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